Brief Communications

Transformation of 2-alkylcyclohexane-1,3-diones into δ-oxo acid esters in an acidic medium*

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Refluxing of a solution of 2,2,5,5-tetramethylcyclohexane-1,3-dione in a dilute H_2SO_4 —ethanol mixture smoothly gave ethyl 3,3,6-trimethyl-5-oxoheptanoate. Under the same conditions, hydrolysis of enamino ketone was followed by analogous ring opening of intermediate 2,4,6-trimethylcyclohexane-1,3-dione.

Key words: cyclohexane-1,3-diones, ring opening, δ -oxo acid esters.

Cyclic 1,3-diketones are known to undergo ring opening under the action of alkalis to give δ -oxo acids (*e.g.*, see Refs 1, 2). However, the possibility of an analogous ring opening in the presence of a protic acid has been reported only in a few papers;^{3–5} in some cases, this reaction has been only postulated when considering a probable mechanism of a more complex process.^{4,5} We found that such a transformation can be effected by treating a substrate with dilute H_2SO_4 in ethanol. For instance, refluxing of a solution of 2,2,5,5-tetramethylcyclohexane-1,3-dione² (1) in 50% H_2SO_4 —EtOH (1:10) gave oxo ester 2 in good yield (Scheme 1).

An analogous transformation accompanied the hydrolysis of racemic 3-amino-2,4(R^*),6(S^*)-trimethylcyclohex-2-en-1-one (3) under the above conditions. Apparently, the reaction proceeds through intermediate 2,4,6-trimethylcyclohexane-1,3-dione (4) to give a mixture of diastereomeric oxo esters 5 (~1:1, 1 H NMR data).

Scheme 1

i. 50% H₂SO₄—aqueous EtOH, refluxing.

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It should be noted that 5,5-dimethylcyclohexane-1,3-dione (dimedone) containing no alkyl substituent at the C(2) atom remains virtually unchanged under these conditions.

Experimental

IR spectra were recorded on a Specord M-80 instrument. 1 H NMR spectra were recorded on a Bruker AC-300 spectrometer in CDCl₃; chemical shifts are referenced to residual signals of the solvent (δ 7.27). The mass spectrum (EI, 70 eV) was recorded on a Finnigan MAT ITD-700 instrument. The $R_{\rm f}$ value is given for a Silufol plate with a fixed SiO₂ layer.

Ethyl 3,3,6-trimethyl-5-oxoheptanoate (2). A solution of diketone 1 (1.68 g, 10.0 mmol) in a mixture of EtOH (30 mL) and 50% H₂SO₄ (6 mL) was refluxed for 23 h and then concentrated *in vacuo*. The residue was treated with MeOBu^t and water. The organic layer was separated, washed with water and brine, dried with Na₂SO₄, and concentrated *in vacuo*. The residue (1.9 g) was distilled to give oxo ester 2 (1.80 g, 84%), b.p. 56-58 °C (1.5 Torr). ¹H NMR (300.13 MHz), δ : 1.03 (d, 6 H, MeC(6), H₃C(7), J = 6.1 Hz); 1.07 (s, 6 H, 2 MeC(3)); 1.23 (t, 3 H, MeCH₂O, J = 6.7 Hz); 2.45 (s, 2 H, H₂C(2)); 2.55 (sept, 1 H, HC(6), J = 6.1 Hz); 2.58 (s, 2 H, H₂C(4)); 4.09 (q, 2 H, CH₂O, J = 6.7 Hz) (*cf.* Ref. 2 for Me ester).

Ethyl 2,4-dimethyl-5-oxoheptanoate (5) (1:1 mixture of diastereomers). Analogously, a solution of enamino ketone 3 (1.69 g, 11.0 mmol) was refluxed for 30 h. The product (1.85 g) was chromatographed on SiO_2 (50 g) with light petroleum—MeOBu^t (4:1) as an eluent to give a mixture of diastereomers 5 (1.50 g, 68%) as a colorless oil, b.p. 50—54 °C (2 Torr), R_f 0.59 (hexane—MeOBu^t, 7:3). Found (%): C, 65.87; H, 10.39. $C_{11}H_{20}O$. Calculated (%): C, 65.97; H, 10.06. MS, m/z:

200 [M]⁺. IR (CHCl₃), v/cm⁻¹: 664, 720, 792, 1016, 1116, 1180, 1296, 1352, 1396, 1460, 1580, 1608, 2860, 2932, 2968, 3004, 3432, 3536. ¹H NMR, δ : 0.97—1.08 (m, 6 H, MeC(2), MeC(4)); 1.11 (br.t, 3 H, HC(7), J = 6.7 Hz); 1.23 (br.t, 3 H, MeCH₂O, J = 7.0 Hz); 1.30 (ddd, 1 H, HC(3), J = 13.8 Hz, J = 7.6 Hz, J = 5.9 Hz)*; 1.57 (ddd, 1 H, HC(3), J = 14.1 Hz, J = 9.0 Hz, J = 4.8 Hz)*; 1.77 (ddd, 1 H, HC(3), J = 14.1 Hz, J = 8.9 Hz, J = 5.3 Hz)*; 2.03 (ddd, 1 H, HC(3), J = 13.9 Hz, J = 8.9 Hz, J = 6.0 Hz)*; 2.25 (m, 4 H, HC(2), HC(4), H₂C(6)); 4.09 (br.q, 2 H, H₂CO, J = 7.0 Hz).

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^{*} Nonoverlapped signals for the protons of the stereoisomers with the relative integral intensity $\sim 1/2$.